A NOVEL APPROACH TO PERIPLANONE-B INVOLVING AN INTRAMOLECULAR DIELS-ALDER REACTION WITH FURAN-DIENE AND ALLENE-DIENOPHILE

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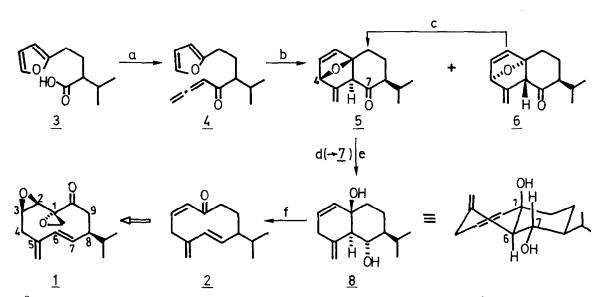
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<u>Abstract</u> - The 6-step sequence to trienone 2, a known intermediate for the total synthesis of periplanone-B, involves the conversion of 3 into 4 using dilithiated propadiene, the IMDA of 4 to <u>exo</u>-adducts 5 and 6, the conversion of 5 to diol 8 via a radical anion promoted cleavage of the oxygen bridge, and a in-situ low temperature Grob fragmentation to 2.

After the discovery in 1952 that females of the species <u>Periplanata americana</u>, the American cockroach, produce a very potent sex excitant,² two extremely active compounds, periplanones -A and -B, were isolated by Persoons, and a germacranoid structure proposed for the latter.³ The unambiguous structural assignment of periplanone-B as <u>1</u> resulted from the pioneering work of Still, that culminated in the first total synthesis of (±)-<u>1</u>.⁴ Later total syntheses of (±)-<u>1</u> were reported by Schreiber,⁵ Hauptmann and Walker,⁶ Takahashi,⁷ and of natural (-)-<u>1</u> by Kitahara.⁸ In these approaches the construction of the 10-membered ring resulted either from an oxyanion-Cope rearrangement^{4,5} or from an alkylative cyclization.⁶⁻⁸

We herein describe an expeditious synthesis of racemic trienone $\underline{2}$, a known intermediate for the synthesis of periplanone-B.^{5,8} The proposed route involves three key-steps : (i) the non-stereoselective IMDA reaction of furan-allene $\underline{4}$, (ii) the conversion of adduct $\underline{5}$ into diol $\underline{8}$, possessing the appropriate functionality and stereochemistry for (iii) eventual Grob fragmentation to trienone $\underline{2}$ (scheme 1). The present synthesis serves to illustrate the synthetic potential of combining both furan-diene and allene-dienophile in a intramolecular Diels-Alder reaction.⁹ Examples of the use of furan as a diene partner in this reaction type have been scarce, despite some obvious synthetic potential.¹⁰ On the other hand, allene-dienophiles are virtually unknown in IMDA reactions,¹¹ especially when furan is involved as an endocyclic diene.^{12,13}

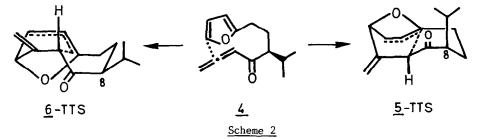
Acid <u>3</u> was obtained via alkylation of the dianion of isovaleric acid (LDA, THF, 45°C, 2 h) with 1-(2-bromoethyl)furan¹⁴ (THF, rt, 12 h; 60 % yield).^{15,16} The direct introduction of the allene moiety was successfully achieved by treatment of the lithium salt of <u>3</u> (<u>n</u>-BuLi, -78°C) with 5 equiv of dilithiated propadiene¹⁷ in THF at rt.¹⁸ Monolithiated propadiene was found unreactive under the same conditions.¹⁹ Without further purification allene $\underline{4}^{16}$ is usually subjected to the cycloaddition.



^a 5 eq Li₂C₃H₂ (from allene : 2 eq <u>n</u>-BuLi, -55°C, 2 h), THF, rt, 10 min; ^b benzene, 80°C, 12 h (65 % from <u>3</u>); ^c mesitylene, 164°C, 24 h to <u>5:6</u>, ratio 2:1; ^d Li, liq. NH₃, NH₄Cl, -78°C (42 %); ^e LiDBB radical anion, THF, -50°C; aniline (68 %); ^f 2 eq <u>n</u>-BuLi, -30°C, 30 min; 3 eq triflic anhydride, ether, -20°C, 15 min (38 %).

Scheme 1

In view of the prior results obtained in this IMDA-type,¹³ the cycloaddition of $\underline{4}$ is expected to proceed irreversibly under normal thermal conditions,²⁰ and to yield adducts of the <u>exo</u>-type (i.e., <u>trans</u>-oriented angular H and oxygen bridge). The transition states leading to the two <u>exo</u>-adducts $\underline{5}$ and $\underline{6}$, in which the asynchronicity of the Diels-Alder reaction is reflected and wherein the pseudo 10-membered ring would adopt a boat-chair-boat conformation, reveal that the eventual outcome of the cycloaddition would be primarily dictated by the orientation of the isopropyl group (scheme 2).²¹ Since force field calculations have shown that there is no relevant difference in conformational energy between an equatorial and axial group at this position (cf. C-8) in a BCB-form,²² the reaction is not expected to be stereoselective. This is borne out by the experiment : both at rt (CH₂Cl₂, 3 days) and in refluxing benzene (12 h; 90 % conversion after 3 h), the cycloaddition of $\underline{4}$ leads to a similar mixture consisting of the two expected <u>exo</u>-adducts $\underline{5}$ and $\underline{6}$,¹⁶ and one <u>endo</u>-adduct with undefined stereochemistry, in a ratio 5:4:1, respectively (65 % combined yield from 3).²³



Under thermodynamic control, the more stable adduct $\underline{5}$ with equatorial isopropyl group is expected to predominate. Not surprisingly, the less stable isomer <u>6</u> was found reluctant to cyclorevert under thermal conditions.²⁰ Reflux in mesitylene (N₂, 164°C) for 24 h was found necessary for obtaining an equilibrium mixture of <u>5</u> and <u>6</u> ratio 2:1, respectively.

The conversion of adduct 5 into diol 8 involves two steps : (i) the reduction of the carbonyl into alcohol 7^{16} with the equatorial orientation at C-7 (Li, liq. NH₃, NH₄Cl, -78°C; 42 % yield next to 45 % starting material); (ii) the selective reductive cleavage of the oxygen bridge at C-4 using lithium di-<u>tert</u>-butylbiphenyl radical anion (THF, -50°C; 68 % of $8.^{24}$ Quite to our surprise, the unsaturated diol 8 could be readily purified by column chromatography on neutral aluminum oxide and HPLC.¹⁶

Derivative <u>8</u> possesses the necessary features for Grob fragmentation to trienone <u>2</u> with the required (E)-6,7-bond, i.e., the antiperiplanar orientation of the eventual leaving group at C-7 to the 1,6-fusion bond of the <u>trans</u>-decalin system.²⁵ Treatment of the dilithium salt of <u>8</u> (<u>n</u>-BuLi, -30°C, ether, 30 min) with trifluoromethanesulfonic anhydride (3 eq) readily led at -20°C to the desired <u>2</u> (44 % isolated after chromatography on aluminum oxide, followed by HPLC), next to starting material (25 %). Comparison of the spectral data of <u>2</u>¹⁶ with those reported for (-)-<u>2</u> fully corroborated its structure.⁸

Although several steps need further optimization, some features of the above sequence, in which no protective groups were used, are noteworthy : the direct introduction of the allene moiety on acid $\underline{3}$ using dilithiated propadiene, the use of lithium radical anion (LiDBB) for effecting the selective cleavage of the oxygen bridge in $\underline{7}$, and, finally, the in situ low temperature Grob fragmentation of $\underline{8}$ into trienone $\underline{2}$. Also, the brievety of the route (6 steps from isovaleric acid) further exemplifies the synthetic potential of the IMDA reaction using the combination of furan-diene and allene-dienophile.

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